# Faculty of Sciences Syllabus for Bachelor of Science (Honours) Physics (Under Continuous Evaluation System)

(SEMESTER: I)

Session – (2020-21)



Kanya Maha Vidyalaya, Jalandhar (Autonomous) The Heritage Institution

# Kanya Maha Vidyalaya, Jalandhar (Autonomous)

# SCHEME AND CURRICULUM OF EXAMINATIONS OF THREE YEAR DEGREE PROGRAMME Bachelor of Science (Honours) Physics Session-2020-21

# Semester-I

Sr. No.	Course Code	Course Type	Course Title	Max Marks				Examin ation
				Total	Ext	Ext		time in
					L	Р	-	Hours)
1.	BOPL-1421 BOPL-1031 BOPL-1431	С	Punjabi(Compulsory <sup>1</sup> Basic Punjabi <sup>2</sup> Punjab History and Culture	50	40	-	10	3
2.	BOPL-1102	С	Communication skills in English	n 50	40	-	10	3
3.	BOPL-1393	С	Mechanics-I	75	60	-	15	3
4.	BOPL-1394	С	Electricity & Magnetism-I	75	60	-	15	3
5.	BOPL-1335	С	Mathematics-I	50	40	-	10	3
6.	BOPL-1086	С	Chemistry-I	50	40	-	10	3
7.	BOPP-1397	С	Physics Lab-I	50	-	40	10	3
8.	BOPP-1088	С	Chemistry Lab-I	50	-	40	10	3
9.	AECD-1161	AC	*Drug Abuse: Problem Management & Prevention (Compulsory)	50	40	-	10	3
10.	SECF-1492	AC	*Foundation Program	25	20		5	1

\*Marks of these papers will not be added in total marks

1 In Lieu of Punjabi (Compulsory) for students from Punjab

2. In Lieu of Punjabi (Compulsory) for students outside Punjab (Other States)

# PROGRAM SPECIFIC OUTCOMES (PSOs)

1. Students will demonstrate proficiency in mathematics and the mathematical concepts needed for a proper understanding of physics.

2. Students will demonstrate knowledge of classical mechanics, electromagnetism, quantum mechanics, Electronics, Nuclear and thermal physics, and be able to apply this knowledge to analyse a variety of physical phenomena.

3. Students will demonstrate knowledge of organic, Physical and inorganic chemistry and are able to recognize and apply the principles of atomic and molecular structure to predict chemical properties and chemical reactivity.

4. Students will show that they have learned laboratory skills, enabling them to take measurements in a physics and chemistry laboratory and analyse the measurements to draw valid conclusions.

5. Upon completion of this degree, students will be able to understand theoretical concepts of instruments that are commonly used in most physics and chemistry fields as well as interpret and use data generated in instrumental analysers.

6. Students will be capable of oral and written scientific communication, and will prove that they can think critically and work independently.

#### SESSION 2020-21 Punjabi (Compulsory)

# BACHELOR OF SCIENCE (HONOURS) PHYSICS SEMESTER-I

# COURSE CODE- BOPL-1421

# COURSE OUTCOMES

Co1:'ਸਾਹਿਤ ਦੇ ਰੰਗ' ਪੁਸਤਕ ਦੇ ਕਵਿਤਾ ਭਾਗ ਨੂੰ ਪੜ੍ਹਾਉਣ ਦਾ ਮਨੋਰਥ ਵਿਦਿਆਰਥੀਆਂ ਅੰਦਰ ਕਵਿਤਾ ਪ੍ਰਤੀ ਦਿਲਚਸਪੀ, ਸੂਝ ਨੂੰ ਪੈਦਾ ਕਰਨਾ ਹੈ ਤਾਂ ਕਿ ਉਹ ਆਧੁਨਿਕ ਦੌਰ ਵਿਚ ਚਲ ਰਹੀਆਂ ਕਾਵਿ ਧਾਰਾਵਾਂ ਅਤੇ ਕਵੀਆਂ ਬਾਰੇ ਗਿਆਨ ਹਾਸਲ ਕਰ ਸਕਣ। ਇਸ ਦਾ ਹੋਰ ਮਨੋਰਥ ਕਵਿਤਾ ਦੀ ਵਿਆਖਿਆ, ਵਿਸ਼ਲੇਸ਼ਣ ਤੇ ਮੁਲੰਕਣ ਦੀ ਪ੍ਰਕਿਰਿਆ ਤੋਂ ਜਾਣੂ ਕਰਾਉਣਾ ਵੀ ਹੈ ਤਾਂ ਕਿ ਉਹ ਸਮਕਾਲੀ ਸਮਾਜ ਦੀਆਂ ਸਮੱਸਿਆਵਾਂ ਨੂੰ ਸਮਝ ਸਕਣ ਅਤੇ ਆਲੋਚਨਾਤਮਕ ਦ੍ਰਿਸ਼ਟੀ ਬਣਾ ਸਕਣ।

Co2:'ਸਾਹਿਤ ਦੇ ਰੰਗ' ਪੁਸਤਕ ਦੇ ਕਹਾਣੀ ਭਾਗ ਨੂੰ ਸਿਲੇਬਸ ਵਿਚ ਸ਼ਾਮਿਲ ਕਰ ਕੇ ਵਿਦਿਆਰਥੀਆਂ ਅੰਦਰ ਕਹਾਣੀ ਪੜ੍ਹਣ ਦੀ ਰੁਚੀ ਨੂੰ ਪੈਦਾ ਕਰਨਾ ਹੈ ਅਤੇ ਕਹਾਣੀ ਜਗਤ ਨਾਲ ਜੋੜਣਾ ਹੈ।

Co3:ਪੈਰ੍ਹਾ ਰਚਨਾ ਅਤੇ ਪੈਰ੍ਹਾ ਪੜ੍ਹ ਕੇ ਪ੍ਰਸ਼ਨਾਂ ਦੇ ਉਤਰ ਦੇਣ ਦਾ ਮਨਰੋਥ ਵਿਦਿਆਰਥੀਆਂ ਦੀ ਬੁੱਧੀ ਨੂੰ ਤੀਖਣ ਕਰਦਿਆਂ ਉਨਾਂ ਦੀ ਲਿਖਣ ਪ੍ਰਤਿਭਾ ਨੂੰ ਉਜਾਗਰ ਕਰਨਾ ਹੈ।

Co4: ਧੁਨੀ ਵਿਉਂਤ ਪੜ੍ਹਣ ਨਾਲ ਵਿਦਿਆਰਥੀ ਧੁਨੀਆਂ ਦੀ ਉਚਾਰਨ ਪ੍ਰਣਾਲੀ ਤੋਂ ਵਾਕਫ਼ ਹੋਣਗੇ।

# SESSION 2020-21 Punjabi (Compulsory)

# BACHELOR OF SCIENCE (HONOURS) PHYSICS SEMESTER-I

# COURSE CODE- BOPL-1421

ਸਮਾਂ : 3 ਘੰਟੇ	Maximum Ma Theory CA	urks: 50 : 40 :10
ਪਾਠ ਕ੍ਰਮ ਅਤੇ ਪਾਠ ਪੁਸਤਕਾਂ ਯੂਨਿਟ-I		
ਸਾਹਿਤ ਦੇ ਰੰਗ (ਸੰਪਾ.ਡਾ ਮਹਿਲ ਸਿੰਘ),ਭਾਗ ਪਹਿਲਾ(ਕਵਿਤਾ ), ਰਵੀ ਸਾਹਿਤ ਪ੍ਰਕਾਸ਼ਨ,ਅੰਮ੍ਰਿਤਸਰ। (ਸਾਰ,ਵਿਸ਼ਾ ਵਸਤੂ) (ਡਾ. ਹਰਿਭਜਨ ਸਿੰਘ,ਪਾਸ਼,ਸੁਰਜੀਤ ਪਾਤਰ ਕਵੀ ਪਾਠਕ੍ਰਮ ਦਾ ਹਿੱਸਾ ਨਹੀਂ ਹਨ)	8 ਅੰਕ	
ਯੂਨਿਟ-II		
ਸਾਹਿਤ ਦੇ ਰੰਗ (ਸੰਪਾ.ਡਾ ਮਹਿਲ ਸਿੰਘ),ਭਾਗ ਪਹਿਲਾ(ਕਹਾਣੀ), ਰਵੀ ਸਾਹਿਤ ਪ੍ਰਕਾਸ਼ਨ,ਅੰਮ੍ਰਿਤਸਰ। (ਸਾਰ,ਵਿਸ਼ਾ ਵਸਤੂ) (ਕੋਈ ਇਕ ਸਵਾਰ,ਘੋਟਣਾ, ਆਪਣਾ ਆਪਣਾ ਹਿੱਸਾ ਕਹਾਣੀਆਂ ਪਾਠਕ੍ਰਮ ਦਾ ਹਿੱਸਾ ਨਹੀਂ ਹਨ)	8 ਅੰਕ	
ਯੂਨਿਟ-III		
ਪੈਰ੍ਹਾ ਰਚਨਾ		
ਪੈਰ੍ਹਾ ਪੜ੍ਹ ਕੇ ਪਸ਼੍ਰਨਾਂ ਦੇ ਉਤਰ।	8 ਅੰਕ	
ਯੂਨਿਟ-IV		
(ੳ) ਪੰਜਾਬੀ ਧੁਨੀ ਵਿਉਂਤ : ਪਰਿਭਾਸ਼ਾ ਤੇ ਉਚਾਰਨ ਅੰਗ		
(ਅ) ਸਵਰ, ਵਿਅੰਜਨ	8 ਅੰਕ	

- 1. ਪ੍ਰਸ਼ਨ ਪੱਤਰ ਦੇ ਚਾਰ ਸੈਕਸ਼ਨ ਹੋਣਗੇ।ਸੈਕਸ਼ਨ A-D ਤੱਕ ਦੇ ਪ੍ਰਸ਼ਨ ਯੂਨਿਟ I-IV ਵਿਚੋਂ ਪੁੱਛੇ ਜਾਣਗੇ। ਹਰ ਸੈਕਸ਼ਨ ਵਿਚ ਦੋ ਪ੍ਰਸ਼ਨ ਪੁੱਛੇ ਜਾਣਗੇ।
- ਵਿਦਿਆਰਥੀ ਨੇ ਕੁਲ ਪੰਜ ਪ੍ਰਸ਼ਨ ਕਰਨੇ ਹਨ। ਹਰ ਸੈਕਸ਼ਨ ਵਿਚੋਂ ਇਕ ਪ੍ਰਸ਼ਨ ਲਾਜ਼ਮੀ ਹੈ। ਪੰਜਵਾਂ ਪ੍ਰਸ਼ਨ ਕਿਸੇ ਵੀ ਸੈਕਸ਼ਨ ਵਿਚੋਂ ਕੀਤਾ ਜਾ ਸਕਦਾ ਹੈ।
- 3. ਹਰੇਕ ਪ੍ਰਸ਼ਨ ਦੇ 08 ਅੰਕ ਹਨ।
- 4. ਪੇਪਰ ਸੈੱਟ ਕਰਨ ਵਾਲਾ ਜੇਕਰ ਚਾਹੇ ਤਾਂ ਪ੍ਰਸ਼ਨਾਂ ਦੀ ਵੰਡ ਅਗੋਂ ਵੱਧ ਤੋਂ ਵੱਧ ਚਾਰ ਉਪ ਪ੍ਰਸ਼ਨਾਂ ਵਿਚ ਕਰ ਸਕਦਾ ਹੈ।

# **BASIC PUNJABI**

# In lieu of Punjabi (Compulsory)

# **COURSE CODE -BOPL-1031**

# **Course outcomes**

CO1:ਮੁੱਢਲੀ ਪੰਜਾਬੀ ਪੜ੍ਹਾਉਣ ਦਾ ਮਨੋਰਥ ਵਿਦਿਆਰਥੀਆਂ ਨੂੰ ਪੰਜਾਬੀ ਭਾਸ਼ਾ ਨੂੰ ਸਿਖਾਉਣ ਦੀ ਪ੍ਰਕਿਰਿਆ ਵਿਚ ਪਾ ਕੇ ਇਕ ਹੋਰ ਭਾਸ਼ਾ ਸਿੱਖਣ ਦਾ ਮੌਕਾ ਪ੍ਰਦਾਨ ਕਰਨਾ ਹੈ।

CO2:ਇਸ ਵਿਚ ਵਿਦਿਆਰਥੀ ਨੂੰ ਬਾਰੀਕਬੀਨੀ ਨਾਲ ਭਾਸ਼ਾ ਦਾ ਅਧਿਐਨ ਕਰਵਾਇਆ ਜਾਵੇਗਾ।

CO3:ਵਿਦਿਆਰਥੀਆਂ ਨੂੰ ਪੰਜਾਬੀ ਸ਼ਬਦ ਰਚਨਾ ਤੋਂ ਜਾਣੂ ਕਰਵਾਇਆ ਜਾਵੇਗਾ।

CO4:ਮੁੱਢਲੀ ਪੰਜਾਬੀ ਪੜ੍ਹਾਉਣ ਦਾ ਮਨੋਰਥ ਵਿਦਿਆਰਥੀਆਂ ਨੂੰ ਨਿੱਤ ਵਰਤੋਂ ਦੀ ਪੰਜਾਬੀ ਸ਼ਬਦਾਵਲੀ ਬਾਰੇ ਦੱਸਣਾ ਹੈ।

CO5:ਮੁੱਢਲੀ ਪੰਜਾਬੀ ਪੜ੍ਹਾਉਣ ਦਾ ਮਨੋਰਥ ਵਿਦਿਆਰਥੀਆਂ ਦਾ ਸ਼ਬਦ ਘੇਰਾ ਵਿਸ਼ਾਲ ਕਰਨਾ ਹੈ।

CO6:ਵਿਦਿਆਰਥੀਆਂ ਨੂੰ ਪੰਜਾਬੀ ਵਿਚ ਹਫ਼ਤੇ ਦੇ ਸੱਤ ਦਿਨਾਂ ਦੇ ਨਾਂ, ਬਾਰ੍ਹਾਂ ਮਹੀਨਿਆਂ ਦੇ ਨਾਂ, ਰੁੱਤਾਂ ਦੇ ਨਾਂ, ਇਕ ਤੋਂ ਸੌ ਤੱਕ ਗਿਣਤੀ ਸ਼ਬਦਾਂ ਵਿਚ ਸਿਖਾਉਣਾ ਹੈ।

# SESSION 2020-21 BASIC PUNJABI

# In lieu of Punjabi (Compulsory)

# **COURSE CODE - BOPL-1031**

ਸਮਾਂ : 3 ਘੰਟੇ	Maximum Marks: 50		
	Theory	: 40	
	СА	: 10	
ਪਾਠ ਕ੍ਰਮ			
ਯੂਨਿਟ-I			
ਪੈਂਤੀ ਅੱਖਰੀ, ਅੱਖਰ ਕ੍ਰਮ, ਪੈਰ ਬਿੰਦੀ ਵਾਲੇ ਵਰਣ ਅਤੇ ਪੈਰ ਵਿਚ ਪੈਣ ਵਾਲੇ	ਓ ਵਰਣ ਅਤੇ ਮਾਤ੍ਰਵਾਂ (ਮੁੱ	ਰੁੱਢਲੀ	
ਜਾਣ ਪਛਾਣ) ਲਗਾਖਰ (ਬਿੰਦੀ, ਟਿੱਪੀ, ਅੱਧਕ) : ਪਛਾਣ ਅਤੇ ਵਰਤੋਂ ।			
		08ਅੰਕ	
ਯੂਨਿਟ-II			
ਪੰਜਾਬੀ ਸ਼ਬਦ ਬਣਤਰ : ਮੁੱਢਲੀ ਜਾਣ ਪਛਾਣ (ਸਾਧਾਰਨ ਸ਼ਬਦ, ਸੰਯੁਕਤ ਸ਼ਬ	ਜ਼ਦ, ਮਿਸ਼ਰਤ ਸ਼ਬਦ, ਮੂ	ਲ	
ਸ਼ਬਦ, ਅਗੇਤਰ ਅਤੇ ਪਿਛੇਤਰ)		08ਅੰਕ	
ਯੂਨਿਟ-III			
ਨਿੱਤ ਵਰਤੋਂ ਦੀ ਪੰਜਾਬੀ ਸ਼ਬਦਾਵਲੀ : ਬਾਜ਼ਾਰ, ਵਪਾਰ, ਰਿਸ਼ਤੇਨਾਤੇ, ਖੇਤੀ ਅ	।ਤੇ ਹੋਰ ਧੰਦਿਆਂ ਆਦਿ	ਨਾਲ	
ਸੰਬੰਧਤ।		08 ਅੰਕ	
ਯੂਨਿਟ-IV			
ਹਫ਼ਤੇ ਦੇ ਸੱਤ ਦਿਨਾਂ ਦੇ ਨਾਂ, ਬਾਰ੍ਹਾਂ ਮਹੀਨਿਆਂ ਦੇ ਨਾਂ, ਰੁੱਤਾਂ ਦੇ ਨਾਂ, ਇਕ ਤੋਂ	ਸੌ ਤਕ ਗਿਣਤੀ ਸ਼ਬਦਾਂ	ਵਿਚ ।	
ਅੰਕ ਵੰਡ ਅਤੇ ਪਰੀਖਿਅਕ ਲਈ ਹਦਾਇਤਾਂ			
1 ਪ੍ਰਸਨ ਪੱਤਰ ਦੇ ਦਾਰ ਸੈਕਸਨ ਹੋਣਗੇ।ਸੈਕਸਨ A-D ਤੱਕ ਦੇ ਪ	ਸਨ ਸ਼ਨਿਟ LIV ਵਿੱ	रें ऑने चाइतो। तन	

1. ਪ੍ਰਸ਼ਨ ਪੱਤਰ ਦੇ ਚਾਰ ਸੈਕਸ਼ਨ ਹੋਣਗੇ।ਸੈਕਸ਼ਨ A-D ਤੱਕ ਦੇ ਪ੍ਰਸ਼ਨ ਯੂਨਿਟ I-IV ਵਿਚੋਂ ਪੁੱਛੇ ਜਾਣਗੇ। ਹਰ ਸੈਕਸ਼ਨ ਵਿਚ ਦੋ ਪ੍ਰਸ਼ਨ ਪੁੱਛੇ ਜਾਣਗੇ।

- ਵਿਦਿਆਰਥੀ ਨੇ ਕੁੱਲ ਪੰਜ ਪ੍ਰਸ਼ਨ ਕਰਨੇ ਹਨ। ਹਰ ਸੈਕਸ਼ਨ ਵਿਚੋਂ ਇਕ ਪ੍ਰਸ਼ਨ ਲਾਜ਼ਮੀ ਹੈ। ਪੰਜਵਾਂ ਪ੍ਰਸ਼ਨ ਕਿਸੇ ਵੀ ਸੈਕਸ਼ਨ ਵਿਚੋਂ ਕੀਤਾ ਜਾ ਸਕਦਾ ਹੈ।
- 3. ਹਰੇਕ ਪ੍ਰਸ਼ਨ ਦੇ 08 ਅੰਕ ਹਨ।
- ਪੇਪਰ ਸੈੱਟ ਕਰਨ ਵਾਲਾ ਜੇਕਰ ਚਾਹੇ ਤਾਂ ਪ੍ਰਸ਼ਨਾਂ ਦੀ ਵੰਡ ਅੱਗੋਂ ਵੱਧ ਤੋਂ ਵੱਧ ਚਾਰ ਉਪ ਪ੍ਰਸ਼ਨਾਂ ਵਿਚ ਕਰ ਸਕਦਾ ਹੈ।

# (Semester-I) Session 2020-21 Course Title: Punjab History and Culture (From Earliest Times to C 320) (Special paper in lieu of Punjabi Compulsory) (For those students who are not domicile of Punjab) Course Code: BOPL-1431

# **COURSE OUTCOMES**

After completing Semester I and course on Punjab History and Culture students of History will be able to identify and have a complete grasp on the sources & writings of Ancient Indian History of Punjab.

**CO1:** Identify and describe the emergence of earliest civilizations in: Indus Valley Civilization and Aryan Societies.

CO2: Identify and analyses the Buddhist, Jain and Hindu faith in the Punjab

**CO3:** Analyses the emergence of Early Aryans and Later Vedic Period, their Society, Culture, Polity and Economy

**CO4:** To make students understand the concepts of two faiths Jainism and Buddhism, its principles and their application and relevance in present times

# (Semester-I) Session 2020-21 Course Title: Punjab History and Culture (From Earliest Times to C 320) (Special paper in lieu of Punjabi Compulsory) (For those students who are not domicile of Punjab) Course Code: BOPL-1431

# **Examination Time: 3 Hours**

Max. Marks: 50 Theory: 40 C A: 10

# **Instructions for the Paper Setters**

- 1. Question paper shall consist of four Units
- 2. Examiner shall set 8 questions in all by selecting **Two Questions** of equal marks from each Unit.
- 3. Candidates shall attempt **5 questions** in **600 words**, by at least selecting **One Question** from each Unit and the **5<sup>th</sup> question** may be attempted from any of the **four Units**.

4.Each question will carry 8 marks.

# Unit-I

- 1. Physical features of the Punjab
- 2. Sources of the ancient history of Punjab

# Unit-II

- 3. Harappan Civilization: social, economic and religious life of the Indus Valley People.
- 4. The Indo-Aryans: Original home

# Unit-III

- 5. Social, Religious and Economic life during Early Vedic Age.
- 6. Social, Religious and Economic life during Later Vedic Age.

# UNIT-IV

- 7. Teachings of Buddhism
- 8. Teachings of Jainism

# **Suggested Readings**

- L. M Joshi (ed.), *History and Culture of the Punjab*, Art-I, Patiala, 1989 (3<sup>rd</sup> edition)
- L.M. Joshi and Fauja Singh (ed.), *History of Punjab*, Vol.I, Patiala 1977.
- Budha Parkash, *Glimpses of Ancient Punjab*, Patiala, 1983.
- B.N. Sharma, Life in Northern India, Delhi. 1966.
- Chopra, P.N., Puri, B.N., & Das, M.N.(1974). A Social, Cultural & Economic History of India, Vol. I, New Delhi: Macmillan India.

#### (Semester I) Session 2020-21 COMMUNICATION SKILLS IN ENGLISH (Theory) Course Code: BOPL -1102

# **COURSE OUTCOMES**

#### At the end of this course, the students will develop the following Skills:

CO 1: Reading skills that will facilitate them to become an efficient reader

CO 2: The ability to realise not only language productivity but also the pleasure of being able to articulate well

CO 3: The power to analyse, interpret and infer the ideas in the text

CO 4: The ability to have a comprehensive understanding of the ideas in the text and enhance their critical thinking

CO 5: Writing skills of students which will make them proficient enough to express ideas in clear and grammatically correct English

CO 6: Ability to plan, organise and present ideas coherently on a given topic

CO 7: The skill to use an appropriate style and format in writing letters (formal and informal)

#### (Semester I) Session 2020-21 COMMUNICATION SKILLS IN ENGLISH (Theory) Course Code: BOPL -1102 Time: 3 Hours Max. Marks: 50 Theory: 40 Continuous Assessment: 10

Instructions for the paper setter and distribution of marks:

The question paper will consist of four sections. The candidate will have to attempt five questions in all selecting one from each section and the fifth question from any of the four sections. Each question will carry 8 marks.

Section-A: Two questions of theoretical nature will be set from Unit I.

Section-B: Two comprehension passages will be given to the students based on Unit II.

Section-C: Two questions will be given from Unit III.

Section-D: Two questions will be set from Unit IV.

 $(8 \times 5 = 40)$ 

(1/3)

#### (Semester I) Session 2020-21 COMMUNICATION SKILLS IN ENGLISH (Theory) Course Code: BOPL -1102

The syllabus is divided in four units as mentioned below:

Unit I

Reading Skills: Reading Tactics and strategies; Reading purposes-kinds of purposes and associated comprehension; Reading for direct meanings.

Unit II

Reading for understanding concepts, details, coherence, logical progression and meanings of phrases/ expressions.

Activities:

- Comprehension questions in multiple choice format
- Short comprehension questions based on content and development of ideas

#### Unit III

Writing Skills: Guidelines for effective writing; writing styles for application, personal letter, official/ business letter.

Activities

- Formatting personal and business letters.
- Organising the details in a sequential order

#### Unit IV

Resume, memo, notices etc.; outline and revision.

#### Activities:

- Converting a biographical note into a sequenced resume or vice-versa
- Ordering and sub-dividing the contents while making notes.
- Writing notices for circulation/ boards

#### (2/3) Recommended Books:

- 1. Oxford Guide to Effective Writing and Speaking by John Seely.
- 2. Business Communication, by Sinha, K.K. Galgotia Publishers, 2003.
- 3. Business Communication by Sethi, A and Adhikari, B., McGraw Hill Education 2009.
- 4. Communication Skills by Raman, M. & S. Sharma, OUP, New Delhi, India (2011).
- 5. English Grammar in Use: A Self Study Reference and Practice Book Intermediate Learners Book

by Raymond Murphy, Cambridge University Press.

# SEMESTER-I

# **MECHANICS-I**

# **Course Code: BOPL-1393**

# **Course Outcomes: Mechanics -I**

After passing this course, students will be able to:

CO1: Understand the various coordinate systems and its applications. Students will be able to know the conservations laws and the symmetries of space & time.

CO2: Know the fundamental forces of nature, concept of center mass, central forces and the motion of particle under central force and to determine the turning points of orbit.

CO3: Understand the frames of reference, coriolis forces and its applications and effect of rotation of earth on g.

CO4: understand the elastic collision in different systems, cross section of elastic scattering as well as Rutherford scattering and know the motion of rigid body.

# BACHELOR OF SCIENCE (HONURS) PHYSICS (SESSION 2020-21) SEMESTER-I COURSE CODE: BOPL-1393 MECHANICS-I

Maximum Marks: 75 (External 60 + Internal 15) Pass Marks: 21 Examination Time: 3 Hours Total Teaching hours: 60

#### **Instructions for the Paper Setters:**

Eight questions of equal marks are to be set, two in each of the four Sections (A-D). Questions of Sections A-D should be set from Units I-IV of the syllabus respectively. Questions may be subdivided into parts (not exceeding four). Candidates are required to attempt five questions, selecting at least one question from each section. The fifth question may be attempted from any Section.

Note: Students can use Non-Scientific calculators or logarithmic tables.

# <u>Unit- I</u>

Co-ordinate system and Motion of a Particle: Cartesian and Spherical polar co-ordinate systems; area, volume, displacement, velocity and acceleration in these systems. Solid angle, Newton's laws of motion. Relationship of conservation laws and symmetries of space and time.

#### <u>Unit- II</u>

Conservation of Momentum and Collisions: Internal forces and momentum conservation, Centre of mass, Elastic collisions in laboratory and centre of mass systems; velocities, angles, energies in these systems and their relationships. Conservation of angular momentum and examples-shape of the galaxy, angular momentum of solar system. Torques due to internal forces, angular momentum about center of mass. Cross-section elastic scattering and impact parameter, Rutherford scattering.

# <u>Unit- III</u>

Inverse-Square-Law Force: Forces in nature (qualitative). Central forces, Potential energy and force between a point mass and spherical shell, a point mass and solid sphere, gravitational and electrostatic self-energy. Two body problem and concept of reduced mass. Motion of a body under central force, equation of orbit in inverse-square force field. Kepler's laws and their derivation.

#### <u>Unit- IV</u>

Dynamics of Rigid Bodies: Equation of motion of a rigid body, rotational motion of a rigid body in general and that of plane lamina. Rotation of angular momentum vector about a fixed axis. Angular momentum and kinetic energy of a rigid body about principal axis, Euler's equations. Precession and elementary gyroscope, Spinning top.

- 1. Mechanics-Berkeley Physics Course, Vol-I by C. Kittel, W. D. Knight, M. A. Ruderman, C. A. Helmholtz and R. J. Moyer-Tata McGraw Hill Publishing Company Ltd., New Delhi.
- 2. Fundamentals of Physics by D. Halliday, R. Resnick and J. Walker -Wiley India Pvt. Ltd., New Delhi.
- 3. Introduction to Classical Mechanics by R. G. Takwale & P.S. Puranik. Tata McGraw Hill Publishing Company Ltd., New Delhi
- 4. An introduction to Mechanics by D. Kleppne r& R. Kolenkow. Tata McGraw Hill Publishing Company Ltd., New Delhi.
- 5. Mechanics by H.S. Hans & S.P Puri, Tata McGraw Hill Publishing Company Ltd. New Delhi.
- 6. Analytical Mechanics by S. K. Gupta, Modern Publishers.

# **ELECTRICITY & MAGNETISM-I**

#### Course No. BOPL-1394

#### **Course Outcomes: Electricity and Magnetism-I**

After passing this course the students will be able to:

- CO1: understand the vector calculus and vector algebra and its applications in electricity and magnetism. The students will be able to solve the electrostatic problems with the help of Gauss law and Coulomb's law.
- CO2: understand the applications of scalar potential for the calculation of electric field and electric potential due to an arbitrary charge distribution.
- CO3: solve the problems with the help of method of images and understand the conduction of electric current and fundamental laws of electricity and relate the electric and magnetic fields in two inertial frames of reference.
- CO4: able to understand electric field, potential and polarization of different media and related quantities.

# BACHELOR OF SCIENCE (HONURS) PHYSICS (SESSION 2020-21) SEMESTER-I COURSE CODE: BOPL-1394 ELECTRICITY AND MAGNETISM-I

Maximum Marks: 75 (External 60 + Internal 15) Pass Marks: 21

Examination Time: 3 Hours Total Teaching hours: 60

# **Instructions for the Paper Setters:**

Eight questions of equal marks are to be set, two in each of the four Sections (A-D). Questions of Sections A-D should be set from Units I-IV of the syllabus respectively. Questions may be subdivided into parts (not exceeding four). Candidates are required to attempt five questions, selecting at least one question from each section. The fifth question may be attempted from any Section.

Note: Students can use Non-Scientific calculators or logarithmic tables.

# <u>Unit- I</u>

<u>Calculus of Vectors</u> : Introduction to gradient, divergence & curl; their physical significance. Rules for vector derivatives, useful relations involving gradient, divergence & curl. Fundamental theorem for gradients, Gauss's and Stoke's theorems.

# <u>Unit- II</u>

**Electrostatics and Electric Current**: Electric charge and its properties, Coulomb's law. The electric field due to a point charge and continuous charge distributions, Field due to electric dipole, Field lines, flux, Gauss's law and its applications. Curl of electric field. Relation between potential and electric field. Poisson's and Laplace's equations. Electric potential due to different charge distribution: Wire, Ring, Disc, Spherical Shell, Sphere, dipole etc. The potential energy for a point and continuous charge distribution.

# <u>Unit- III</u>

**Field of Moving Charges**:- Conductors in the electrostatic field, Capacitors, Current and current density, drift velocity, expression for current density vector, equation of continuity. Ohm's Law and expression for electrical conductivity, limitations of Ohm's law. Equipotential surface method of electrical images, Measurement of charge in motion, Transformation of electric and magnetic fields in different frames of references, Electric field due to moving charges, electric force in two inertial frames, Interaction between moving charges.

#### Unit- IV

**Electric Fields in Matter**: Moments of charge distribution, Potential and field of a dipole, torque and force on a dipole in an external electric field, polarizability tensor, Electric field caused by polarized matter, Electric field of Polarized Sphere, Dielectric sphere in a uniform electric field, Field of a charge in a dielectric medium, Electric susceptibility and atomic polarizability tensor, Polarization in alternating fields.

- 1. Introduction to Electrodynamics by D.J. Griffiths, Perason Prentice Hall of India, New Delhi
- 2. Electricity & Magnetism by E.M. Purcell, Berkeley Physics Course Vol. 2, McGraw Hill, New York
- 3. Fundamental of Physics by D. Halliday, R. Resnick and J. Walker (6th edition)-John Wiley India Pvt. Ltd.,.
- 4. Electricity and Magnetism by A. S. Mahajan & A. A. Rang Wala, Tata –McGraw Hill Publication Pvt. Ltd.

# B.Sc. (Hons.) Physics Semester–I Course Title: Mathematics-I Course Code: BOPL-1335

# **Course outcomes**

After the completion of this course, students should be able to :

**CO 1**: Give argument related to limits, continuity and derivative of a function.

**CO 2:** Understand the concept of maxima and minima of a function of a single variable.

**CO 3:** Explain the significance of Roll's theorem, Mean Value theorem, and Taylor's and Maclaurin's theorem to find the expansions of functions.

**CO 4:** Demonstrate the geometrical meaning of integral calculus as an area and their uses in the determination of C.G & moments of inertia.

**CO 5:**Understand how to solve linear differential equations with constant coefficients and linear homogeneous and inhomogeneous differential equations of second order.

# BACHELOR OF SCIENCE (HONURS) PHYSICS (SESSION 2020-21) SEMESTER-I COURSE CODE: BOPL-1335 MATHEMATICS-I

Maximum Marks: 50 (External 40 + Internal 10) E Pass Marks: 18 T

# Examination Time: 3 Hours Total Teaching hours: 60

# **Instructions for the Paper Setters:**

Eight questions of equal marks are to be set, two in each of the four Sections (A-D). Questions of Sections A-D should be set from Units I-IV of the syllabus respectively. Questions may be subdivided into parts (not exceeding four). Candidates are required to attempt five questions, selecting at least one question from each section. The fifth question may be attempted from any Section.

Note: Students can use Non-Scientific calculators or logarithmic tables.

# UNIT –I

Functions and Derivatives: Limits, continuity and derivative of function. Differentiation of standard functions, Successive differentiation. Geometrical significance of derivative. Maxima and Minima of a function of single variable. Partial differentiation. Chain rule of differentiation. 15 Lectures

# UNIT –II

Differential Calculus: Statement of Rolle's theorem and Mean value theorem, Taylor's and Maclaurins theorems and their applications to formal expansion of functions. Tangents and normals. Basic idea about asymptotes.

15 Lectures

# UNIT –III

Integral Calculus: Integration as inverse of differentiation. Indefinite integrals of standard forms. Method of substitution. Integration using partial fractions. Integration by parts. Reduction formulae. Definite integrals. Definite integral as limit of a sum and geometrical interpretation as an area.

15 Lectures

# UNIT –IV

Differential Equations: Definition & formation of differential equations. Linear differential equation of first order and first degree. Linear homogenous and inhomogeneous differential equation of second order. Linear differential equations with constant coefficients.

15 Lectures

Text Book: Higher Engineering Mathematics: B.S. Grewal, Delhi, Khanna

Reference Books:

- 1. Differential Calculus: Shanti Narayan, New Delhi, Shyam Lal
- 2. Integral Calculus: Shanti Narayan, Delhi, S. Chand
- 3. Mathematical Hand Book: M. Vygodsky, Mir, Moscow

# Bachelor of Science (Honours) Physics Semester–I Session: 2020-21 Course Title: Chemistry I Course Code: BOPL-1086

# **Course outcomes:**

Students will be able to

CO1: differentiate between chiral and achiral compounds, configuration and conformation.

CO2: understand the concept of isomerism

CO3: understand the resolution of enantiomers and differentiate between dextrorotatory and laevorotatory compounds.

CO4: do conformational analysis of ethane, butane, cyclohexane, monosubstituted and disubstituted cyclohexane.

CO5: explain the various methods of formation and chemical reactions of alkanes, alkenes and alkynes.

CO6: understand functional group transformation by nucleophilic substitution.

CO7: describe the mechanism and stereochemistry of nucleophilic substitution reactions.

CO8: understand the principles of nucleophilic addition to carbonyl groups.

# Bachelor of Science (Honours) Physics Semester–I Session: 2020-21 Course Title: Chemistry I Course Code: BOPL-1086

#### **Examination Time: 3 Hours**

Max. Marks: 50 Theory: 40, CA: 10

#### **Instructions for the Paper Setters:**

Eight questions of equal marks (Specified in the syllabus) are to be set, two in each of the four Sections (A-D). Questions may be subdivided into parts (not exceeding four). Candidates are required to attempt five questions, selecting at least one question from each Section. The fifth question may be attempted from any Section.

#### Note: Students can use Non-Scientific calculators or logarithmic tables.

# UNIT I

**Stereochemistry**: Molecular chirality, enantiomers/symmetry in achiral structures, chiral centres in chiral molecules, properties of chiral molecules-optical activity, absolute and relative configuration, the Cahn-Ingold Prelog R-S notional system physical properties of enantiomers.

#### **UNIT II**

**Chemistry of alkanes and alkenes:** Conformations of alkanes and cycloalkanes: conformational analysis of ethane, butane, cyclohexane, mono substituted and disubstituted cyclohexane. Stereochemistry of alkenes, naming stereo isometric alkenes by the E-Z system, mechanism of hydrogenation of alkenes, stereochemistry of hydrogenation of cyclo alkenes, Dehydration of alcohols and region selectivity of these reactions. Acid catalysed dehydration of alcohols with complete mechanistic discussion, Mechanism of dehydro halogenation of alkyl halides (E mechanism), stereo selective and anti-elimination in E reactions, the E Mechanism, electrophilic addition of hydrogen halides to alkenes its region selectivity explained on the basis of mechanism, free radical addition of hydrogen bromide to alkenes, acid catalysed hydration of alkene with mechanism stereochemistry of halogen addition to alkenes and its mechanistic explanation. Hypohalous acid addition to alkenes, epoxidation of alkenes.

**Alkynes:** Acidity of acetylene and terminal alkenes, metal ammonia reduction of alkyne, addition of hydrogen halides and water to alkynes, with detailed discussion of mechanism of these reactions, the Diels Alder reaction.

#### **UNIT-III**

Nucleophilic substitution and addition reaction: Functional group transformation by nucleophilic substitution, the biomolecular ( $SN^2$ ), mechanism of nucleophilic substitution, stereochemistry of  $SN^2$  reactions, how  $SN^2$  reactions occur, steric effect in  $SN^2$  reactions, nucleophiles and nucleophilicity, the unimolecular ( $SN^1$ ) mechanism of nucleophilic substitution, carbocation stability and the rate of

substitution, by the  $SN^1$  mechanism stereochemistry of  $SN^1$  reactions, carbocation real arrangements in  $SN^1$  reactions, solvent effects, substitution and elimination as competing reactions.

# **UNIT-IV**

**Principles of nucleophillic addition to carbonyl groups:** Hydration acetal formation, cyanohydrin formation; reactions with primary and secondary amines, Whittig reaction, mechanism of halogenation, acid and base catalysed enolization, haloform reaction, aldol condensation, conjugate nucleophillic addition to unsaturated carbonyl compounds

# **Text and Reference Books:**

- 1. R.T. Morison and R.N. Boyd, Organic Chemistry.
- 2. I.L. Finar, Organic Chemistry, Vol. I IV ed.
- 3. Advanced Organic Chemistry, Reactions Mechanisms and Structure by J. March.
- 4. Schaum's Outlines Series Theory and Problems of Organic Chemistry by Herbert Meislick and Jacob Sharefkin
- 5. Problems and their solution in Organic chemistry by I.L. Finar, Modern Organic Chemistry by J.D. Robbert and M.C. Caserio.
- 6. Organic Chemistry by D.J. Cram and G.S. Hammond.
- 7. J.E. Banks, Naming Organic Compounds Programmed Introduction to Organic Chemistry.
- 8. E.L. Eliel, Stereochemistry of carbon compounds.
- 9. A. Carey, Organic Chemistry.

# **Course Outcomes: Physics Lab-I Sem I**

# Course No. BOPP-1397

CO1: Students will be able to find the value of acceleration due to gravity using pendulums.

- CO2: It will give understanding of collisions in 1-Dimension.
- CO3: It helps to study the moment of inertia of a body & on what factors it depends.

# BACHELOR OF SCIENCE (HONURS) PHYSICS (SESSION 2020-21) SEMESTER-I COURSE CODE: BOPL-1397 PHYSICS LAB-I

Maximum Marks: 50 (External 40 + Internal 10) Pass Marks: 18 Examination Time: 3 Hours Total Teaching hours: 90

# **Instructions to Practical Examiner**

Question paper is to be set on the spot jointly by the external and internal examiners. Two copies of the same to be submitted for the record to COE office, KanyaMahaVidyalaya, Jalandhar

**General Guidelines for Practical Examination** 

I. The distribution of marks is as follows:

i) One experiment 20 Marks

ii) Brief Theory 6 Marks

iii) Viva–Voce 7 Marks

iv) Record (Practical file) 7Marks

**II.** There will be one sessions of 3 hours duration. The paper will have one session and will consist of 8 experiments out of which an examinee will mark 6 experiments and one of these is to be allotted by the external examiner.

III. Number of candidates in a group for practical examination should not exceed 20.

IV. In a single group no experiment be allotted to more than three examinee in any group.

#### List of experiments-

- 1. To measure the time periods of oscillation for the objects of various geometrical shapes but of same mass.
- 2. To study rotational motion using a flywheel and hence show that torque is proportional to angular acceleration.
- 3. To find the moment of inertia of an irregular body about an axis through its centre of gravity with a torsion pendulum.
- 4. To determine the moment of inertia of a flywheel.
- 5. To determine the Young's modulus by bending.
- 6. Determination of Poisson's ratio for rubber.
- 7. To verify laws of conservation of (a) linear momentum, (b) kinetic energy in elastic collisions using one dimensional collisions of hanging spheres. (c) Also determine energy transfer and coefficient of restitution.
- 8. To determine modulus of rigidity of copper wire by Maxwell needle experiment.
- 9. To determine the coefficient of viscosity of glycerine by Stoke's method.
- 10. To find the unknown capacitance of a capacitor by flashing and quenching
- 11. of a neon lamp.
- 12. Measurement of capacitance, determination of permittivity of a medium air and relative permittivity by de–Sauty's bridge.

13. To study the variation in resistance of filament of a bulb with temperature. **Reference Books:** 

1. Practical Physics, C.L. Arora, S. Chand & Co.

# Bachelor of Science (Honours) Physics Semester–I Session: 2020-21 Course Title: Chemistry Lab-I Course Code: BOPL-1088

# **Course outcomes**

Students will be able to

CO1: develop skills required for the qualitative analysis of organic compounds,

CO2: detect elements (N, S and halogens)

CO3: detect functional groups (phenolic, carboxylic, carbonyl, esters, carbohydrates, amines, amides, nitro) in simple organic compounds

CO4: determine the physical constants of organic compounds.

CO5: prepare the derivatives of organic compounds.

#### Bachelor of Science (Honours) Physics Semester-I

Session: 2020-21

Course Title: Chemistry Lab-I

Course Code: BOPL-1088

Examination Time: 3 Hours

Max. Marks: 50

Practical: 40, CA: 10

**Instruction for practical examiner**: Question paper is to be set on the spot jointly by the Internal and External Examiners. Two copies of the same should be submitted for the record to COE Office, Kanya Maha Vidyalaya, Jalandhar.

# **General Guidelines for Practical Examination**

The preliminary examination of physical and chemical characteristics (physical state, colour, odor and ignition tests), elemental analysis (nitrogen, sulphur, chlorine, bromine, iodine), solubility tests including acid-base reactions, classification tests involving functional reactivity other than acid-base test, preparation of derivatives for given pure organic compounds.

The following categories of compounds should be analysed:

-Phenols

- Carboxylic acids
- -Carbonyl compounds (ketones, aldehydes)
- -Carbohydrates
- -Aromatic amines
- -Amides and Nitro compounds

# Suggested Book:

- 1. Practical Organic Chemistry by F.G. Mann and B.C. Saunders
- 2. Practical Organic Chemistry by Vogel

# SEMESTER – I DRUG ABUSE Course Code: AECD-1161 (Theory) Course Outcomes:

- CO 1. This information can include factual data about what substance abuse is; warning signs of addiction; information about how alcohol and specific drugs affect the mind and body;
- CO 2. How to be supportive during the detoxification and rehabilitation process.
- CO 3. Main focus of substance abuse education is teaching individuals about drug and alcohol abuse and how to avoid, stop, or get help for substance use disorders.
- CO 4. Substance abuse education is important for students alike; there are many misconceptions about commonly used legal and illegal substances, such as alcohol and marijuana.

# SEMESTER – I DRUG ABUSE Course Code: AECD-1161

(Theory)

# Max.Marks: 50 CA: 10

Time: 3 Hrs Theory: 40

#### Instructions for the Paper Setter

Eight questions of equal marks are to be set, two in each of the four Sections (A-D). Questions of Sections A-D should be set from Units I-IV of the syllabus respectively. Questions may be subdivided into parts (not exceeding four). Candidates are required to attempt five questions, selecting at least one question from each section. The fifth question may be attempted from any Section.

#### UNIT-I

**1)Meaning of Drug Abuse:** Concept and Overview, Historical Perspective of Drug Abuse, Drug Dependence, Drug Addiction, Physical and Psychological Dependence: Drug Tolerance and withdrawal symptoms.

# UNIT-II

#### 2) Types of Abused Drugs and their Effects -I

1) Stimulants: Amphetamines – Benzedrine, Dexedrine, Cocaine.

2) Depressants: Alcohol Barbiturates: Nembutal, Seconal, Phenobarbital and Rohypnol.

3) Narcotics: Heroin, Morphine, Oxycodone.

# UNIT-III

# 3) Types of abused drugs and their effects - II

1) Hallucinogens: Cannabis, Marijuana, Hashish, Hash Oil, MDMA, LSD.

2) Steroids.

# UNIT-IV

**4) Nature and Extent of the Problem:** Magnitude or prevalence of the menace of Drug Abuse in India and Punjab, Vulnerable groups by age, gender and economic status, Signs and Symptoms of Drug Abuse: Physical, Academic, Behavioural and Psychological Indicators.

# **References:**

1. Ahuja, Ram (2003), Social Problems in India, Rawat Publication, Jaipur.

2. Extent, Pattern and Trend of Drug Use in India, Ministry of Social Justice and Empowerment, Government of India, 2004.

3. Inciardi, J.A. 1981. The Drug Crime Connection. Beverly Hills: Sage Publications.

4. Kapoor. T. (1985) Drug epidemic among Indian Youth, New Delhi: Mittal Pub.

5. Modi, Ishwar and Modi, Shalini (1997) *Drugs: Addiction and Prevention*, Jaipur: Rawat Publication.

6. National Household Survey of Alcohol and Drug abuse. (2003) New Delhi, Clinical Epidemiological Unit, All India Institute of Medical Sciences, 2004.

7. Sain, Bhim 1991, *Drug Addiction Alcoholism*, Smoking obscenity New Delhi: Mittal Publications.

8. Sandhu, Ranvinder Singh, 2009, *Drug Addiction in Punjab*: A Sociological Study. Amritsar: Guru Nanak Dev University.

9. Singh, Chandra Paul 2000. *Alcohol and Dependence among Industrial Workers*: Delhi: Shipra.

10. Sussman, S and Ames, S.L. (2008). *Drug Abuse: Concepts, Prevention and Cessation,* Cambridge University Press.

# FOUNDATION PROGRAMME

# Course Title: Foundation Programme Course Duration: 30 hours

# Course intended for: Semester I students of undergraduate degree programmes of all

# streams. Course Credits: 1 Course Code: SECF-I

# **PURPOSE & AIM**

This course has been designed to strengthen the intellectual foundation of all the new entrants in the college. One of the most common factors found in the students seeking admission in college after high school is the lack of an overall view of human history, knowledge of global issues, peaks of human intellect, social/political benchmarks and inventors & discoverers who have impacted human life. For a student, the process of transformation from school to college is full of apprehension and intimidation of the system. The Foundation Programme intends to bridge the gap between high school and college education and develop an intellectual readiness and base for acquiring higher education.

# **INSTRUCTIONAL OBJECTIVES**

• to enable the students to realise their position in the whole saga of time and space

• to inculcate in them an appreciation of life, cultures and people across the globe

• to promote, in the students, an awareness of human intellectual history

• to make them responsible and humane world citizens so that they can carry forward the rich legacy of humanity

#### FOUNDATION PROGRAMME

# **Course title: Foundation Programme Course duration: 30 hours** Course intended for: Sem. I students of all streams (UG Only) Course credits: 1 **Course code: SECF-I**

#### CURRICULUM

Course Code: V1	Course Credits: 1 Total Contac	tact Hours: 20	
MODULE	TITLE	HOURS	
Ι	Introduction & Initial Assessment	2	
II	The Human Story	3	
III	The Vedas, The Gita& Eastern Philosophy	2.5	
IV	The Holy Bible& Genesis	2.5	
V	Woman: A Journey through the Ages	2.5	
VI	Changing Paradigms in Society, Religion & Literature	2.5	
VII	Makers of Modern India	2.5	
VIII	Racism & Martin Luther King Jr.	2.5	
IX	Modern India at a Glance: Political & Economic Perspective	2.5	
Х	Technology & Human Life	2.5	
XI	The KMV Experience	2.5	
XII	Final Assessment, Feedback & Closure	2.5	

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#### **EXAMINATION**

- Total Marks: 25 (Final Exam: 20; Internal Assessment: 5)
- Final Exam: multiple choice quiz. Marks 20; Time: 1 hour
- Internal Assessment: 5 (Assessment: 3; Attendance:2)

Comparative assessment questions (medium length) in the beginning and close of the programme. Marks: 3; Time: 0.5 hour each at the beginning and end.

- Total marks: 25 converted to grade for final result
- Grading system: 90% marks & above: A grade

80% - 89% marks : B grade 70% - 79% marks : C grade 60% - 69% marks : D grade 50% - 59% marks : E grade Below 50% marks : F grade (Fail - must give the exam again) **SYLLABUS** Module I Being a Human: Introduction & Initial Assessment

- Introduction to the programme
- Initial Assessment of the students through written answers to a couple of questions

# Module 2 The Human Story

- Comprehensive overview of human intellectual growth right from the birth of human history
- The wisdom of the Ancients
- Dark Middle Ages
- Revolutionary Renaissance
- Progressive modern times

• Most momentous turning points, inventions and discoveries

# Module 3 The Vedas, The Gita & The Indian Philosophy

- Origin, teachings and significance of The Vedas
- Upnishads and Puranas
- Karma Theory of The Bhagwad Gita
- Main tenets of Buddhism & Jainism
- Teachings of Guru Granth Sahib

# Module 4 The Holy Bible & Genesis

- Book of Genesis: Creation and Fall
- Noah's Ark
- Moses & The Ten Commandments
- Christ and His teachings
- Christianity and the world

# Module 5 Changing Paradigms in Society, Religion & Literature

- Renaissance: The Age of Rebirth
- Transformation in human thought
- Importance of humanism
- Geocentricism to heliocentricism
- Copernicus, Galileo, Columbus, Darwin and Saint Joan
- Empathy and Compassion

#### Module 6 Woman: A Journey through the Ages

- Status of women in pre-vedic times
- Women in ancient Greek and Roman civilizations
- Women in vedic and ancient India
- Status of women in the Muslim world
- Women in the modern world
- Crimes against women
- Women labour workforce participation
- Women in politics
- Status of women- our dream

# Module 7 Makers of Modern India

- Early engagement of foreigners with India
- Education: The first step to modernization
- Railways: The lifeline of India
- Raja Ram Mohan Roy, Gandhi, Nehru, Vivekanand, Sardar Patel etc.
- Indira Gandhi, Mother Teresa, Homai Vyarawala etc.
- The Way Ahead

# Module 8 Racism: Story of the West

• European beginnings of racism

- Racism in the USA Jim Crow Laws
- Martin Luther King Jr. and the battle against racism
- Apartheid and Nelson Mandela
- Changing face of racism in the modern world

#### Module 9 Modern World at A Glance: Political & Economic Perspective

- Changing world order
- World War I & II
- UNO and The Commonwealth
- Nuclear Powers; Terrorism
- Economic Scenario: IMF, World Bank
- International Regional Economic Integration

# Module 10 Technology and Human Life

- Impact of technology on modern life
- Technological gadgets and their role in our lives
- Technology and environment
- Consumerism and materialism
- Psychological and emotional consequences of technology
- Harmonising technology with ethics and humaneness

# Module 11 The KMV Experience

- Historical Legacy of KMV
- Pioneering role in women emancipation and empowerment
- KMV Contribution in the Indian Freedom Struggle
- Moral, cultural and intellectual heritage of KMV
- Landmark achievements
- Innovative initiatives; international endeavours
- Vision, mission and focus
- Conduct guidelines for students

#### Module 12 Final Assessment, Feedback & Closure

- Final multiple choice quiz
- Assessment through the same questions asked in the beginning
- Feedback about the programme from the students
- Closure of the programme

#### PRESCRIBED READING

• The Human Story published by Dawn Publications